

Conformational Flexibility Determines Selectivity and Antibacterial, Antiplasmodial, and Anticancer Potency of Cationic α -Helical Peptides^{*[5]}

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Background: Antimicrobial peptides (AMPs) have the potential to act against multiple pathogenic targets.

Results: AMPs that maintain conformational flexibility are more potent against multiple pathogens and less hemolytic.

Conclusion: Antimicrobial action and hemolysis proceed via differing mechanisms.

Significance: The potency, selectivity, and ability of AMPs to reach intracellular pathogens can be modulated using general principles.

We used a combination of fluorescence, circular dichroism (CD), and NMR spectroscopies in conjunction with size exclusion chromatography to help rationalize the relative antibacterial, antiplasmodial, and cytotoxic activities of a series of proline-free and proline-containing model antimicrobial peptides (AMPs) in terms of their structural properties. When compared with proline-free analogs, proline-containing peptides had greater activity against Gram-negative bacteria, two mammalian cancer cell lines, and intraerythrocytic *Plasmodium falciparum*, which they were capable of killing without causing hemolysis. In contrast, incorporation of proline did not have a consistent effect on peptide activity against *Mycobacterium tuberculosis*. In membrane-mimicking environments, structures with high α -helix content were adopted by both proline-free and proline-containing peptides. In solution, AMPs generally adopted disordered structures unless their sequences comprised more hydrophobic amino acids or until coordinating phosphate ions were added. Proline-containing peptides resisted ordering induced by either method. The roles of the

angle subtended by positively charged amino acids and the positioning of the proline residues were also investigated. Careful positioning of proline residues in AMP sequences is required to enable the peptide to resist ordering and maintain optimal antibacterial activity, whereas varying the angle subtended by positively charged amino acids can attenuate hemolytic potential albeit with a modest reduction in potency. Maintaining conformational flexibility improves AMP potency and selectivity toward bacterial, plasmodial, and cancerous cells while enabling the targeting of intracellular pathogens.

Linear cationic amphipathic α -helical peptides comprise a class of molecules that often have highly potent antimicrobial properties and consequently have attracted considerable attention in the development of therapeutic agents (1–4). A major goal in this field is to improve the potency of the antimicrobial peptides (AMPs)² while reducing their toxicity to host cells.

The hydrophobicity and secondary structure of linear cationic amphipathic antimicrobial peptides influence their ability to both self-associate and insert into biological membranes. These properties contribute both to antimicrobial potency and host cell toxicity. Many α -helical AMPs incorporate proline or glycine residues, which may enable helical sections to be interrupted, conferring conformational flexibility on the peptide. The α -helix conformation adopted by many AMPs in biological membranes, which constitute either their target of action or the

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^[5] This article contains supplemental Figs. 1–10 and Tables 1–3.

The atomic coordinates and structure factors (codes 2I96, 2I99, and 2I9a) have been deposited in the Protein Data Bank, Research Collaboratory for Structural Bioinformatics, Rutgers University, New Brunswick, NJ (<http://www.rcsb.org/>).

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² The abbreviations used are: AMP, antimicrobial peptide; MIC₅₀, peptide concentration required to inhibit growth by 50%; HC₅₀, peptide concentration required to lyse 50% of red blood cells; EC₅₀, peptide concentration required to kill 50% of parasites or cancer cells.

main barrier to intracellular targets (5), is now regarded as a key determinant of activity (6). Therefore, the presence of helix-destabilizing residues such as glycine or proline would be expected to have a critical effect. Accordingly, a number of previous studies have found that incorporating proline residues into AMPs decreases the ability of the peptide to permeabilize the cytoplasmic membrane of *Escherichia coli* (7) with a concomitant reduction in the antibacterial activity (7–9). Similarly, substituting helix-promoting residues for glycines in the primary sequence of magainin peptides improved the antimicrobial activity considerably, although a small increase in hemolysis was also detected (10).

In contrast, other studies have found that incorporating either L- or D-proline into either the hydrophobic or hydrophilic face of an amphipathic cationic α -helix model peptide, thus disordering the peptide secondary structure, can lead to an improvement, rather than a reduction, in antibacterial activity as well as a reduction in hemolysis (11, 12). This has been ascribed to the ability of proline-containing peptides to resist self-association and interact selectively with anionic lipids as found in the bacterial target membranes but not in those of erythrocytes.

Here, to reconcile these differing views of the effects of proline-induced conformational flexibility on antibacterial potency, we tested the hypothesis that the ability to maintain a disordered structure in certain environments is crucial to AMP potency against a range of bacterial and eukaryotic pathogens and considered the importance of the positioning of the proline residue. We performed a detailed study of the structure and conformation of a series of proline-free and proline-containing model peptides. We found that the properties conferred on AMPs by proline residues depend strongly on the properties of the proline-free template peptide as well as the positioning of the proline residue in the primary sequence. Using circular dichroism (CD) spectroscopy supported by fluorescence spectroscopy, size exclusion chromatography, and NMR diffusion measurements, we show that in solution, analogously to melittin (13), ordering of structure, which may be related to peptide self-association, of model peptides can be induced through the addition of phosphate anion at fixed pH or through increasing the hydrophobicity of the peptides by incorporating phenylalanine residues at the N terminus. Proline-containing peptides resisted this process whether induced by either phosphate anions or increased hydrophobicity. The relationship between the structural ordering of the peptides and their activities against eukaryotic cell targets was notable. Proline-containing peptides were more active against both mammalian cancer cells and the malaria parasite *Plasmodium falciparum* but had substantially reduced hemolytic potential and differential effects against *Mycobacterium tuberculosis*.

Some small but significant differences in activity against *E. coli* and *Pseudomonas aeruginosa* were observed within the series of proline-containing peptides and were ascribed to the effects of the differing positions of the proline residues. By investigating this further, we obtained high resolution structures of three proline-containing analogs in the presence of the anionic detergent sodium dodecyl sulfate (SDS) using NMR spectroscopic methods. The structures reveal that the position

of the single proline in the primary sequence of the model peptide has a considerable effect on the ability of the peptide to adopt an α -helix conformation in a membrane-mimicking environment.

MATERIALS AND METHODS

Peptides—Peptides comprising L-amino acids (Table 1) were purchased from either EZBiolab (Carmel, IN) or Pepceuticals Ltd. (Nottingham, UK) as desalted grade. Peptides comprising D-amino acids were synthesized using standard manual Fmoc (N-(9-fluorenyl)methoxycarbonyl) solid state chemistry. HPLC purification was performed using methanol/water or acetonitrile/water gradients, and the identity of the product was confirmed by matrix-assisted laser desorption ionization mass spectrometry. Peptides were lyophilized from 10% acetic acid to remove the trifluoroacetic acid counterion.

Broth Microdilution Assay—The activities of the peptides against two strains of *E. coli* were assessed in planktonic suspension in polypropylene 96-well plates (Greiner Bio-One, Frickhausen, Germany) according to a modified broth dilution assay (14). *E. coli* NCTC 9001, *P. aeruginosa* PAO1, and *E. coli* TOP10 were gifts from K. D. Bruce and C. Junkes. *E. coli* NCTC 9001, *P. aeruginosa* PAO1, or competent *E. coli* TOP10 were grown without shaking in 50 ml of Mueller-Hinton broth at 37 °C. Peptides were tested in duplicates with two rows allocated for each peptide. In each of columns 2–11, 50 μ l of Mueller-Hinton broth was added under sterile conditions. In the first row, 50 μ l of 256 μ g/ml stock peptide solutions prepared in distilled water were added, and then the broth from the second row was pipetted into the first row and thoroughly mixed before being deposited again in the second row. This process was repeated throughout the tray, providing a 2-fold dilution of peptide with each row. Bacteria with an A_{620} of 0.001 were then added to each well in volumes of 50 μ l, giving a further 2-fold dilution and a final volume of 100 μ l/well. The final column was used either as sterility control (100 μ l of broth) or negative control (no peptide). Plates were incubated overnight at 37 °C, and the A_{620} was read. Growth curves prepared from duplicates were fitted to determine the peptide concentration required to inhibit growth by 50% (MIC₅₀). The MIC₅₀ quoted for each peptide (Table 2) is an average value from at least two independent repeats.

Antituberculosis Assay—The activities of the D-amino acid peptides against *M. tuberculosis* H37Ra, an attenuated *M. tuberculosis* strain commonly used as a standard strain for anti-tuberculosis drug testing, were tested in a manner similar to the broth microdilution assay described above. Duplicate serial dilutions from 100 to 0.78 μ M peptide were prepared in a total volume of 180 μ l of Middlebrook 7H9 broth supplemented with Middlebrook oleic albumin dextrose catalase growth supplement in 96-well plates. A bacterial suspension (20 μ l of a 1×10^6 cfu/ml suspension) was added to each well, giving a final volume of 200 μ l/well. The plates were sealed and incubated at 37 °C with 5% CO₂ for between 4 and 5 weeks prior to inspection. For negative and 1% growth controls, wells contained 20 μ l of a 1×10^6 or 1×10^4 cfu/ml suspension, respectively, and 180 μ l of broth.

Antiplasmodial Assay—The activities of the peptides comprising D-amino acids were tested against both the 3D7 and C10 strains of *P. falciparum* using fluorescent nucleic acid binding dye assays modified from Bacon *et al.* (15). Ring stage parasites maintained in AlbuMAX® RPMI 1640 complete medium were presynchronized using magnetic cell sorting and differential lysis with sorbitol. 2-Fold serial dilutions of each peptide stock in culture medium of concentrations between 64 and 0.064 μM were distributed into either polypropylene 96-well plates (3D7; Greiner Bio-One) or polystyrene Nunc MicroWell™ plates (C10) in a total volume of 50 μL . 150 μL of a thoroughly mixed *P. falciparum* culture at ring stage was added into each well to a final volume of 200 μL , giving final peptide concentrations ranging from 16 to 0.016 μM . The parasitemia was between 1.5 and 2%, whereas the final hematocrit was between 3.5 and 4%. Conditions were prepared for the SYBR Green II assay in triplicate, whereas two additional wells were used for slide preparation with Giemsa stain and determination of parasitemia under a light microscope. Wells containing parasite culture (1.5–2% parasitemia, 3.5–4% final hematocrit) without peptide served as a growth control. For time 0 controls, 50 μL of fresh culture medium and 150 μL of malaria culture were added to a separate 96-well plate, which was immediately stored at -20°C . The plates were gassed with 5% O_2 , 5% CO_2 , and 90% N_2 in a sealed gas chamber for 5 min and then incubated at 37°C for a total of 48 h. Immediately following incubation, blood smears were prepared from each condition, and the plates were stored at -20°C . The assay plates as well as the time 0 plate were thawed at room temperature for the SYBR Green assay. After mixing, 100 μL of the contents of each well was transferred to black 96-well plates, and 100 μL of SYBR Green II/lysis buffer solution (0.2 μL of SYBR Green II/ml of lysis buffer consisting of 20 mM Tris, pH 7.5, 5 mM EDTA, 0.008% (w/v) saponin, and 0.08% (v/v) Triton X-100) was added to each well. The plates were incubated in the dark for 1 h and inspected with a FLUOstar Omega (BMG LABTECH GmbH, Ortenberg, Germany) fluorescence plate reader with excitation at 485 nm and emission monitored at 530 nm.

Hemolysis Assay—A duplicate dilution series of peptides from 256 to 0.016 μM was prepared in 10 mM Tris, 150 mM NaCl, pH 7.4 buffer. Fresh erythrocytes were washed thoroughly in Tris buffer and diluted to get a final concentration of $\sim 2.5 \times 10^9/\text{ml}$. 50 μL of this red blood cell (RBC) suspension was added to 450 μL of the prepared peptide dilution series in microcentrifuge tubes, giving a final RBC concentration of $\sim 2.5 \times 10^8/\text{ml}$. The tubes were incubated for 5 min on ice, then 30 min at 37°C , and a further 5 min on ice before being centrifuged (5 min at $2000 \times g$ at 4°C). 230 μL of supernatant from each tube was transferred to a polystyrene Nunc MicroWell plate. To generate a positive control, 230 μL of a 0.1% Triton X-100 solution was added to erythrocyte pellets prepared from tubes challenged with 450 μL of Tris buffer only. The resulting suspensions (230 μL) were transferred to the same plate, which was read at 540 nm. The percentage of hemolysis was calculated relative to the positive control (100%).

3-(4,5-Dimethylthiazol-2-yl)-2,5-diphenyltetrazolium Bromide Assay—Cytotoxicity was evaluated by performing the 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide

(Sigma) assay (16). 10,000 adenocarcinomic human alveolar basal epithelial cells (A549) or 10,000 RAW 264.7 Abelson murine leukemia virus-transformed macrophages (RAW 264.7) per well were plated in 96-well plates (Costar). The cells were incubated with peptide solutions prepared in serum-free Opti-MEM for 4 h at 37°C . Untreated cells were used as a negative control. The cells were then incubated with 200 μL of prewarmed 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide solution (0.8 mg/ml) per well for a further 2 h. The toxicities of the peptides to both mouse monocytic macrophage cell line RAW 264.7 and human lung cancer cell line A549 at different concentrations were evaluated by comparing the absorbance of formazan crystals (dissolved using isopropanol) formed by the 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide reagent at 570 nm.

Circular Dichroism—Simultaneous UV absorption and CD spectra were acquired on a Chirascan or Chirascan Plus spectrometer (Applied Photophysics, Leatherhead, UK). The instruments were flushed with pure nitrogen gas throughout the measurements. Far-UV spectra were recorded from 260 to 185 nm with a 1-nm spectral bandwidth, 0.5-nm step size, and 1-s spectrometer time per point. A rectangular 0.5-mm pathlength was used. Peptides were dissolved in 5 mM Tris-amine buffer at pH 7.2 to a final concentration of around 20 μM . Peptide samples were also prepared in the presence of 50 mM SDS or 50% (v/v) trifluoroethanol. Peptide titration with sodium phosphate was carried out *in situ* (0.5-mm pathlength) using a 1 M sodium phosphate (pH 7.2)-containing 20 μM peptide as stock solution. Unless otherwise stated, all spectra were measured at 37°C . During data processing, a spectrum of the peptide-free medium or solution was buffer-subtracted, and Savitsky-Golay smoothing with a convolution width of 5 points was applied. CD spectra were normalized for concentration and pathlength and expressed in terms of molar ellipticity per residue. Secondary structure analyses were performed using CDPPro (17).

Tryptophan Fluorescence—Fluorescence emission spectra of tryptophan-containing peptides were acquired using a Cary Eclipse fluorescence spectrophotometer using an excitation wavelength of 280 nm and scanning from 300 to 450 nm at 37°C . Peptides were made up to a final concentration of 20 μM using 5 mM Tris buffer at pH 7.0. 900 μL of a peptide solution was titrated with small volumes of phosphate buffer in a 4×10 -mm cuvette. Next, normalized fluorescence intensities were plotted against wavelength and fitted to a log normal distribution as described (18) in Origin 8 (OriginLab Corp., Northampton, MA), and the wavelengths of the maximum intensities (λ_{max}) were plotted against the concentration of hydrogen phosphate ions.

Size Exclusion Chromatography—75 μL of peptide solutions prepared in the appropriate buffer at a final concentration of 1 mg/ml was injected onto a Superdex™ 75 10/300 GL column (GE Healthcare). Buffers contained 10 mM Tris-amine, 150 mM NaCl, and sodium phosphate at varying concentrations (either 0, 50, 100, 200, or 500 mM) and were adjusted to pH 7.0. Peptides were eluted at a flow rate of 0.5 ml/min on an Agilent 1100 HPLC system with detection at 215 and 280 nm.

Diffusion-ordered Spectroscopy—Samples were prepared in an analogous manner to those used in CD measurements but at

a peptide concentration of 100 μM (i) in 5 mM Tris-amine buffer at pH 7.2 and (ii) titrated to 500 mM sodium phosphate at pH 7.2 with each buffer containing 5% D_2O . NMR spectra were acquired at 298 K on a Bruker Avance 500-MHz spectrometer (Bruker, Coventry, UK) equipped with a cryoprobe. The ^1H 90° pulse length was calibrated for each sample and was found to be close to 7.8 μs for i and 15.6 μs for ii. Diffusion measurements were made using a double stimulated echo pulse program including bipolar gradient pulses and a longitudinal eddy current delay. The spectral width was set to 12 ppm, and the number of scans was set to 32. The relaxation delay was 1 s, and the gradient pulse strength was increased from 5 to 95% of the maximum gradient strength (50 gauss $\cdot\text{cm}^{-1}$). The diffusion time was set to 120 ms, and bipolar gradient pulses of 1.25 ms were applied. Data were processed, and diffusion coefficients were extracted by fitting intensities in the manufacturer's software (TopSpin).

NMR Structure Determination—The NMR samples consisted of a 1 mM peptide solution also containing 100 mM SDS- d_{25} with 5 mM Tris(hydroxymethyl- d_3)-amino- d_2 -methane buffer at pH 7. 10% D_2O -containing trimethylsilyl propanoic acid was added for the lock signal and as an internal chemical shift reference. The temperature was kept constant at 310 K during the NMR experiments. NMR spectra were acquired on a Bruker Avance 500-MHz spectrometer (Bruker) equipped with a cryoprobe. Standard Bruker total correlation spectroscopy and NOESY pulse sequences were used with water suppression using a WATERGATE 3-9-19 sequence with gradients (mlevgpph19 and noesygpph19). The ^1H 90° pulse was calibrated at 37.04 kHz. The total correlation spectroscopy mixing time was 90 ms, and the mixing time for the NOESY spectra was set to 150 ms. The relaxation delay was 1 s. 2048 data points were recorded in the direct dimension, and either 256 or 512 data points were recorded in the indirect dimension. The spectra were processed using Bruker TopSpin. The free induction decay was multiplied by a shifted sine² window function. After Fourier transformation, the spectra were phase-corrected, a base-line correction was applied, and spectra were calibrated to the trimethylsilyl propanoic acid signal at 0 ppm.

Assignments were carried out with SPARKY software (19), and structure calculations were done with ARIA software (20) using NOE restraints of backbone protons and proline H^δ protons only and no dihedral angle restraints. We chose to add the H^δ protons of proline to compensate for the fact that this amino acid residue does not have an NH proton on the backbone and because proline H^δ atoms are easy to assign and do not overlap with other peaks. Restraints involving the side chain atoms of tryptophan were removed because they were violated in every model during and after the structure calculation. The side chain atoms of other residues were removed because the overlapping peaks from especially lysine and leucine residues made it difficult to unambiguously assign and reliably integrate these peaks. The ARIA software was configured to use manual assignments and not allowed to change them. Proton frequency windows were set to 0.02 and 0.04 for the direct and indirect dimensions, respectively. Upper and lower bound corrections were disabled. Structures were calculated using torsion angle molecular dynamics with the slow cooling protocol as published (21) fol-

lowed by refinement in water using default ARIA settings. 100 structures were calculated in each of eight iterations, and the 10 best structures were kept. Network anchoring was enabled during the first three iterations. Results were analyzed with AQUA and PROCHECK_NMR software (22) and python scripts developed in our laboratory. An overview of all backbone NOESY contacts that were used in the structure calculations is available in supplemental Table 3. Structures of LAK160-P7, LAK160-P10, and LAK160-P12 were deposited in the RCSB Protein Data Bank under accession codes 2I96, 2I99, and 2I9a, respectively.

RESULTS

Peptide Design—Peptides (Table 1) were designed to adopt amphipathic α -helix conformations in the appropriate environments. The peptides comprise a series of alternating alanine and leucine residues with two lysine residues located close to each of the N and C termini. The C terminus in each peptide is amidated, which together with the four lysines that are distributed elsewhere in the sequences and describe the “charge angle” when the peptide adopts an amphipathic α -helix confers a nominal charge of +9 to each peptide. We have previously discussed the importance of considering the design of peptides where the angle subtended by the charged residues is varied (23) and recall previous work that describes the activities of cationic α -helical peptides where the hydrophobic moment is either balanced (24) or ignored (25) when altering the charge angle. In the present study, we were interested in the ability of proline residues to modulate self-association while considering their effects on the charge angle when adopting α -helical conformations. Consequently, proline-free and proline-containing peptides (Table 1) were designed without attempting to balance the hydrophobic moment.

A number of previous studies (23, 26–28) have noted the enhanced activities of peptides comprising D-amino acids when compared with analogous peptides prepared from L-amino acids, in particular against *P. falciparum*. To pursue this further, we prepared two series of cationic amphipathic peptides comprising all D-amino acids (supplemental Table 1 and Table 1). The first of these (supplemental Table 1; D-LAKn) was used to identify which charge angle conferred optimal antibacterial activity. Data obtained for the two *E. coli* strains, *P. aeruginosa*, and *M. tuberculosis* (supplemental Table 2 and supplemental Fig. 1) indicated that a charge angle of 120° conferred the most potent antibacterial activity, and this peptide was used as a template for the second series (Table 1; D-LAK120, etc.). In this series, the peptide hydrophobicity was reduced through altering the ratio of alanine to leucine residues (D-LAK120-A) and/or incorporating histidine residues (D-LAK120-H). Proline residues were included in place of leucine or histidine in three peptides at position 13. The final two peptides in this series reassessed the biological effect of varying the charge angle in a proline-containing peptide (D-LAKn-HP13).

In light of the conflicting reports regarding the effect of proline incorporation on antibacterial activity, we investigated whether the positioning of proline would affect activity. For reasons of cost, all L-amino acid peptides were used in this part of the study. Three proline-free peptide templates were selected that are defined by the charge angle, 80°, 120°, or 160°,

TABLE 1

Comparison of physical features of peptides used in this study

Hydrophobicity (H) and mean hydrophobic moment (μH) are shown according to the combined consensus scale or Eisenberg scale and were calculated using the HydroMCalc Java applet made available by Alex Tossi. All peptides contain eight lysine residues and are amidated at the C terminus, which confers a nominal charge of +9 at neutral pH. D-Amino acid residues are shown in italics. Significant changes to the sequences are highlighted: histidine, phenylalanine, and proline residues are highlighted in bold with proline also underlined.

Peptide	Sequence	H^a	H^b	$\mu H^{a,c}$
D-LAK120	<i>KKLALLALKKWLLALKKLALLALKK-NH₂</i>	1.26	−0.05	1.28
D-LAK120-P13	<i>KKLALLALKKWLPALKKLALLALKK-NH₂</i>	0.87	−0.07	1.66
D-LAK120-A	<i>KKLALALAKKWLLALAKKLALALAKK-NH₂</i>	−0.02	−0.08	2.28
D-LAK120-AP13	<i>KKLALALAKKWLPALKKLALALAKK-NH₂</i>	0.00	−0.10	2.25
D-LAK120-H	<i>KKLALHALKKWLHALKKLAHLALKK-NH₂</i>	−0.35	−0.16	2.80
D-LAK120-HP13	<i>KKALAHALKKWLPALKKLALALAKK-NH₂</i>	−1.07	−0.17	3.81
D-LAK80-HP13	<i>KKALAKALKHWLPALHKLAKALAKK-NH₂</i>	−1.07	−0.17	4.02
D-LAK160-HP13	<i>KKALKHALAKWLPALKALAHKLAKK-NH₂</i>	−1.07	−0.17	3.40
LAK80	<i>KKLAKALKLLALLWLKLAKALKKA-NH₂</i>	0.46	−0.09	3.73
LAK80-P7	<i>KKLAKAPKLLALLWLKLAKALKKA-NH₂</i>	0.05	−0.11	3.46
LAK80-P10	<i>KKLAKALKLPALLWLKLAKALKKA-NH₂</i>	0.05	−0.11	3.33
LAK80-P12	<i>KKLAKALKLAPLWLKLAKALKKA-NH₂</i>	0.05	−0.11	4.14
LAK120	<i>KKLALALKKLALLWKLALALAKKA-NH₂</i>	0.46	−0.09	3.02
LAK120-P7	<i>KKLALAPKKLALLWKLALALAKKA-NH₂</i>	0.05	−0.11	2.76
LAK120-P10	<i>KKLALALKKPALLWKLALALAKKA-NH₂</i>	0.05	−0.11	2.62
LAK120-P12	<i>KKLALALKKLAPLWKLALALAKKA-NH₂</i>	0.05	−0.11	3.43
LAK160	<i>KKLALAKLALLWKALALKKKA-NH₂</i>	0.46	−0.09	2.59
LAK160-P7	<i>KKLKLAPAKLALLWKALALKKKA-NH₂</i>	0.05	−0.11	2.34
LAK160-P10	<i>KKLKLALAKPALLWKALALKKKA-NH₂</i>	0.05	−0.11	2.18
LAK160-P12	<i>KKLKLALAKLAPLWKALALKKKA-NH₂</i>	0.05	−0.11	3.00
LAK80-F1	<i>FFKLAKALKLLALLALKLAKALKKA-NH₂</i>	0.41	−0.06	3.36
LAK80-F2	<i>FFKLAKALKLLALLALKLAKALKKA-NH₂</i>	0.78	−0.04	3.55
LAK80-F2-P9	<i>FFKLAKAPKLLALLALKLAKALKKA-NH₂</i>	0.40	−0.06	3.30

^a Combined consensus scale.

^b Eisenberg scale. ^c Mean hydrophobic moment assuming formation of ideal α -helix.

subtended by the four lysine residues described above. For each template peptide, proline residues were substituted for leucine residues in one of three positions. The proline residues were expected to confer conformational flexibility on the peptides while interrupting the α -helix conformation. A range of proline locations was required so that the effects of proline location on the conformation adopted in N- and C-terminal segments of differing lengths could be probed.

Finally, three further peptides comprising all L-amino acids where additions of hydrophobic phenylalanine residues were made at the N terminus were designed. The hydrophobic phenylalanine residues were included to promote peptide self-association because in addition to electrostatic effects aromatic effects are known to contribute to peptide aggregation and fibrillation (29). This allows the effect of proline on self-association driven by hydrophobicity rather than phosphate anion concentration to be tested. Either one or two phenylalanine residues were added to the LAK80 or LAK80-P7 sequences, generating either LAK80-F1 or LAK80-F2 and LAK80-F2-P9 respectively.

Biological Activities of All D-Amino Acid AMPs—Based on the initial screen of D-amino acid peptides according to their charge angle, D-LAK120 was selected as the template for a series of peptides (Table 1) where the effects of reducing peptide hydrophobicity and incorporating proline were probed. The activities of these peptides against Gram-negative bacteria, *M. tuberculosis* H37Ra, two strains of *P. falciparum*, and two mammalian cancer cell lines were determined as were their hemolytic potential (Table 2 and Fig. 1). All of the peptides tested were highly potent against both *E. coli* and *P. aeruginosa* and compare very favorably with the activities of naturally occurring AMPs assayed under the same conditions (30). However, no significant differences in activity were observed as a result of any of the modifications. When tested against *P. falciparum*,

modest but significant ($p < 0.05$) improvements in antiplasmodial activity were sometimes observed in proline-containing peptides when compared with the proline-free analog. Importantly, the antiplasmodial activities need to be compared with the peptide concentration required to lyse 50% of red blood cells (HC₅₀) hemolysis data obtained for each peptide as these reveal whether plasmodial toxicity is specific or occurs only as collateral damage associated with infected erythrocyte lysis. HC₅₀ values greater than 10 times the peptide concentration required to kill 50% of parasites or cancer cells (EC₅₀) for antiplasmodial activity for either strain are highlighted in bold (Table 2) and identify only three peptides that are capable of specific antiplasmodial activity. Light microscopy of erythrocyte cultures of *P. falciparum* (supplemental Fig. 2) shows the effects of one such specific peptide, D-LAK120-AP13, where a dramatic reduction in parasitemia is effected without any observable damage to the erythrocyte hosts. Notably, incorporation of proline in the more hydrophilic D-LAK120-A or D-LAK120-H to give D-LAK120-AP13 or D-LAK120-HP13, respectively, led to dramatic reductions in hemolytic potential, whereas antiplasmodial activity was maintained or improved. This effect was not observed for the most hydrophobic analogs (D-LAK120/D-LAK120-P13). The incorporation of proline in a suitably hydrophilic peptide therefore confers considerable selectivity. Interestingly, altering the charge angle in the D-LAKn-HP13 series also had profound effects on hemolytic potential and hence antiplasmodial specificity. All of the peptides were toxic to the two mammalian cancer cell lines at low micromolar concentrations. However, modest increases in toxicity were again observed for proline-containing peptides when compared with their proline-free analogs.

The antibacterial activity of the peptides against *M. tuberculosis* did not follow the pattern observed for the other targets (Fig. 1). Incorporation of proline in D-LAK120-A or D-LAK120

TABLE 2
Comparison of MIC₅₀, EC₅₀, and HC₅₀ values (μM) for eight designed linear cationic peptides comprising all D-amino acids
Results are an average of two or more independently repeated experiments. HC₅₀ values greater than 10 times the EC₅₀ for antiplasmodial activity for either strain are highlighted in bold. ND, not determined.

Peptide	MIC ₅₀			EC ₅₀				Hemolysis, HC ₅₀
	<i>E. coli</i> NCTC 9001	<i>P. aeruginosa</i> PAO1	<i>E. coli</i> TOP10	<i>P. falciparum</i> 3D7	<i>P. falciparum</i> C10	A549	RAW264.7	
D-LAK120	1.48 ± 0.53	2.71 ± 1.07	0.62 ± 0.15	1.95 ± 0.08	ND	ND	ND	4.06 ± 1.30
D-LAK120-P13	1.54 ± 0.97	1.42 ± 0.41	0.58 ± 0.18	0.88 ± 0.08	2.18 ± 0.41	2.44 ± 0.43	3.80 ± 0.34	11.66 ± 0.03
D-LAK120-A	1.23 ± 0.47	1.83 ± 0.63	0.38 ± 0.21	1.45 ± 0.12	2.13 ± 0.14	5.23 ± 0.35	7.31 ± 2.13	17.42 ± 2.17
D-LAK120-AP13	1.08 ± 0.62	1.22 ± 0.19	0.29 ± 0.09	0.88 ± 0.06	2.36 ± 0.10	3.11 ± 0.57	3.42 ± 0.16	58.61 ± 10.41
D-LAK120-H	1.54 ± 0.01	1.35 ± 0.07	0.40 ± 0.01	1.26 ± 0.14	4.43 ± 0.33	4.54 ± 0.30	5.21 ± 0.57	20.29 ± 2.42
D-LAK120-HP13	ND	ND	ND	ND	2.88 ± 0.25	2.21 ± 0.63	2.33 ± 0.30	79.61 ± 2.80
D-LAK80-HP13	ND	ND	ND	ND	0.96 ± 0.07	6.50 ± 0.37	7.08 ± 1.42	8.61 ± 0.35
D-LAK160-HP13	ND	ND	ND	ND	3.94 ± 0.33	7.58 ± 0.42	3.95 ± 0.40	185.0 ± 18.65

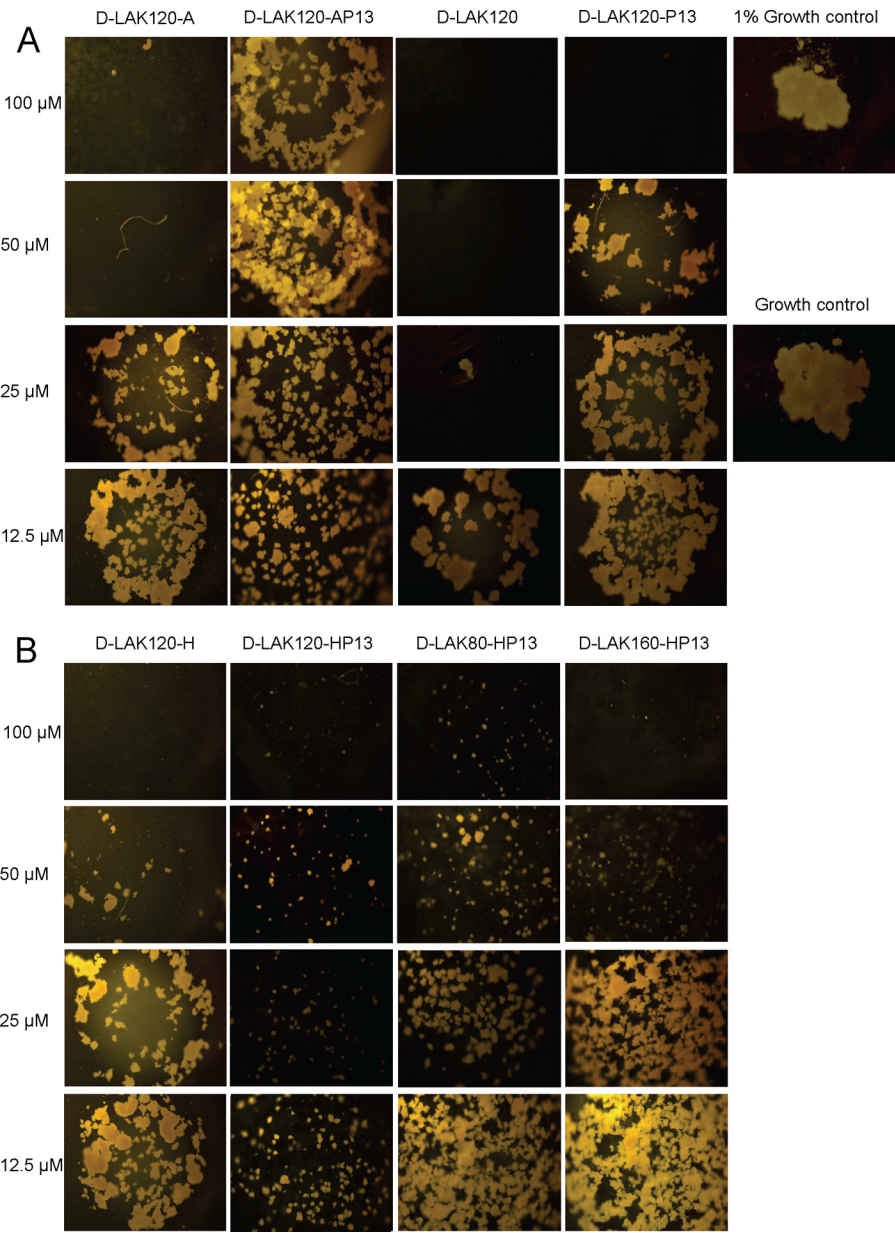


FIGURE 1. The ability of eight D-LAK peptides to inhibit the growth of *M. tuberculosis* H37Ra is shown and compared with a positive growth control (1×10^5 cfu/ml) and a 1% growth control (1×10^4 cfu/ml). When histidine-free peptides are compared (A), a reduction of antibacterial activity is observed when proline is incorporated. For histidine-containing peptides (B), incorporation of proline increases antibacterial activity. Altering the angle subtended by the lysine residues to either 80° or 160° diminishes peptide potency.

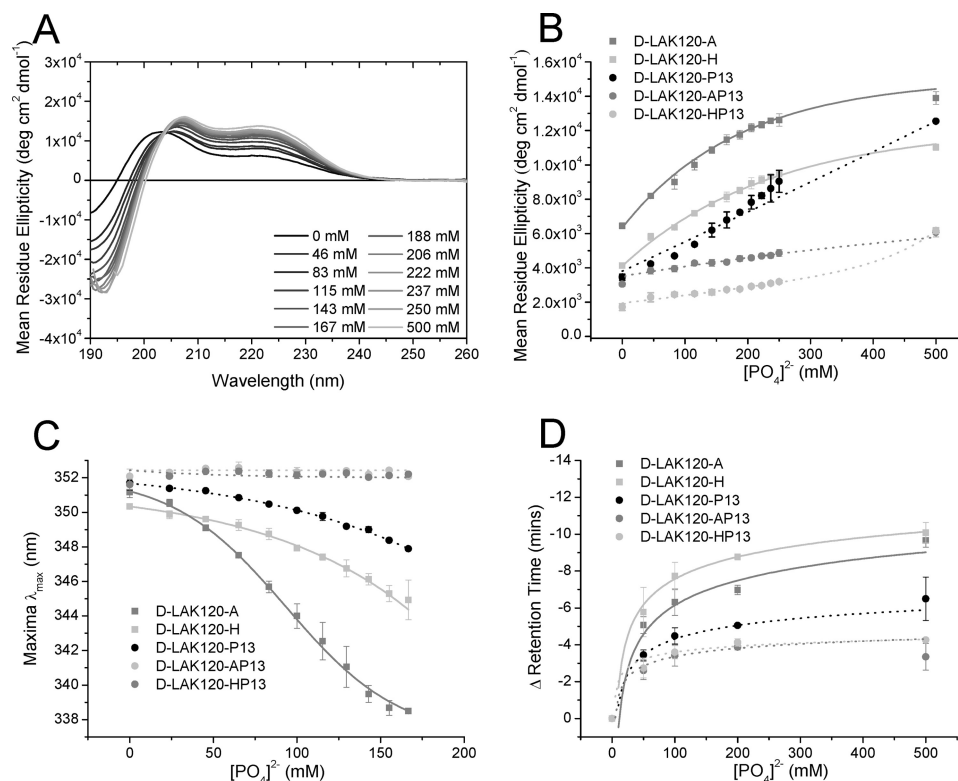


FIGURE 2. Shown are circular dichroism spectra of D-LAK120-A (A) in 5 mM Tris buffer solution, pH 7.3 titrated with increasing concentrations of phosphate buffer at 37 °C. The changes in left-handed α -helix content as monitored by the ellipticity at 220 nm are shown as a function of phosphate concentration (B) and compared with corresponding experiments performed with a further four peptides. The shift in the tryptophan emission maximum (C) and the retention times on a size exclusion column (D) were also evaluated as a function of phosphate concentration and are related to the buildup of secondary structure. Lines are to guide the eye. deg, degrees. Error bars are the standard deviation of two independently repeated experiments.

caused a substantial reduction in efficacy (Fig. 1A). This contrasted with the substantial improvement seen for D-LAK120-H when proline was incorporated (Fig. 1B). The resulting peptide, D-LAK120-HP13, had the highest activity against *M. tuberculosis* of all the peptides tested in the present study. Altering the charge angle in the D-LAKn-HP13 series had a detrimental effect on activity against *M. tuberculosis* (Fig. 1B).

Ordering and Self-association of Peptides—CD measurements of the peptides in solution or in the presence of membrane-mimetic medium provide information on the average secondary structure adopted by each peptide in each differing environment. The conformational change in response to the addition of phosphate buffer was observed for both all D- and all L-amino acid peptide families dissolved in aqueous solution (Fig. 2, A and B, and supplemental Figs. 2–4). Spectra were recorded between 260 and 185 nm with absorbance spectra recorded simultaneously. The phosphate anion absorbs at shorter wavelengths to interfere with the CD signal. Hence, with increasing phosphate concentration, data points at shorter wavelengths are removed according to the monitored absorbance. As shown previously for melittin, a membrane-lytic α -helical peptide from bee venom, phosphate or other coordinating anions can promote self-association of peptides containing arginine or lysine residues by both minimizing electrostatic repulsion and providing cross-link sites where two lysines may simultaneously hydrogen bond with one orthophosphate “bridging” moiety (13).

The comparison of the left-handed α -helix-inducing effect of phosphate anion on proline-free and proline-containing peptides comprising D-amino acids was made by following the development of the positive band at 220 nm, which is characteristic of left-handed α -helix conformation (Fig. 2, A and B). Those peptides containing a proline residue resisted the increase in left-handed α -helix conformation that was characteristic of the proline-free analogs (Fig. 2B). Furthermore, an effect of hydrophobicity was seen with the most hydrophobic proline-containing peptide, D-LAK120-P13, which responded with a greater increase in α -helix conformation at increasing phosphate anion concentration when compared with the more hydrophilic analogs. As with melittin, this increase in α -helix content may be expected to be linked to peptide self-association. Support for this comes from measurements of intrinsic tryptophan fluorescence performed in parallel to the CD studies (Fig. 2C) as well as size exclusion chromatography (Fig. 2D) and diffusion-ordered spectroscopy NMR (supplemental Fig. 3). Fluorescence emission spectra, resulting from the single tryptophan residue at position 11, were monitored following the addition of increasing amounts of phosphate. Blue shifts and intensity reductions in the fluorescence emission maximum were observed that are consistent with the tryptophan residues in individual peptides moving to a more hydrophobic environment as would be encountered in an oligomer and encountering tryptophan residues from other peptides in the oligomer in close proximity, leading to self-quenching (31).

Peptide bonds from neighboring peptides may also contribute weakly to the observed reduction in intensity (32). Qualitatively, the fluorescence data mirror the CD data with the greatest changes in fluorescence and ellipticity at 220 nm in response to phosphate anion occurring for the same proline-free peptides and the least change observed for proline-containing peptides, suggesting that these two parameters are closely linked. To further investigate whether peptide self-association is responsible for the observed conformational and fluorescence emission changes, size exclusion chromatography and diffusion-ordered spectroscopy NMR studies were performed in parallel. The change in retention time on a Superdex 75 10/300 column is shown as a function of phosphate anion concentration (Fig. 2D). Larger particles eluted more rapidly, and with increasing phosphate concentrations, proline-free peptides were observed to elute at increasingly shorter time intervals. This trend was more effectively resisted for proline-containing peptides and is very similar to the effects observed using either CD or fluorescence emission spectroscopy, suggesting that these three techniques are reporting on the same phenomenon. When diffusion-ordered spectroscopy NMR was performed for two pairs of proline-free/proline-containing peptides (supplemental Fig. 3), the results indicated a much better correlation between peptide diffusion coefficients and absolute rather than relative changes in size exclusion retention time. Taken together, these data suggest that the changes in secondary structure and shielding of tryptophan from an aqueous environment are closely related to a relative change in particle size. However, when set against the diffusion coefficients and considering the absolute retention times, it is apparent that the particles comprising disordered proline-containing peptides are larger than those comprising structurally ordered proline-free peptides at low phosphate anion concentrations. Whether this corresponds to comprising a larger number of peptides or reflects a less dense packing is unclear at present. Nonetheless, both size exclusion chromatography and NMR diffusion measurements are in agreement that the presence of proline affects the supramolecular arrangement of the peptides in solution.

All L-amino acid-containing LAK peptides dissolved in Tris/Cl⁻ buffer adopted mostly disordered conformations, whereas addition of phosphate anions induced either small or more dramatic increases in right-handed α -helix conformation content depending on the charge angle and the presence or absence of proline in the primary sequence. For proline-free peptides, LAK80 (supplemental Fig. 4A) and LAK120 (supplemental Fig. 5A) in the absence of phosphate, disordered conformations were adopted that nevertheless indicate equilibrium with an α -helix conformation with notable positive bands at $\sim$ 190 nm. Conformational equilibria have been demonstrated for other linear peptides (33). Addition of phosphate anion produced a substantial increase in α -helix conformation content with the appearance of strong negative bands at 207 and 220 nm and an increase in the positive band intensity at 190 nm until the effects of phosphate ion absorbance obscure this region. In contrast, the proline-free peptide with a greater charge angle, LAK160 (supplemental Fig. 6A), adopted an α -helix-disordered state with much lower α -helix content compared with the two

peptides designed to incorporate more acute charge angles. As with the D-amino acid peptides, the incorporation of proline in either the LAK80 or LAK120 sequences further reduced the α -helix content of the disordered state adopted in Tris buffer and enabled the peptide to resist the ordering effects induced by the addition of phosphate anions (supplemental Figs. 4 and 5). The effect of phosphate anion on the LAK160 peptides (supplemental Fig. 6) reveals that, within this series, both proline-free and proline-containing peptides resist ordering in solution.

To further probe the relationship between structural ordering and the supramolecular organization of LAK peptides in solution and to investigate the ability of proline residues to interfere with this, further variants of LAK80 and its proline-containing analog LAK80-P7 were studied (Fig. 3, A and B). All of these variants, which incorporate either a further one or two phenylalanine residues at the C terminus, adopted conformations with a similarly large α -helix content in the presence of model dimyristoyl phosphatidylcholine/dimyristoyl phosphatidylglycerol (75:25) membranes (Fig. 3B). In contrast, in aqueous solution in the absence of phosphate ions, the secondary structure of the four analogs differed considerably (Fig. 3A). LAK80 adopted a mostly disordered conformation, whereas addition of first one and then two hydrophobic phenylalanine residues to the sequence caused a more modest and then a substantial increase, respectively, in α -helix content. The ability of proline to disrupt α -helix conformation induced by hydrophobic interactions is now highlighted by the disordered conformation adopted by LAK80-F2-P9, a peptide incorporating a single proline residue in addition to the two hydrophobic phenylalanine residues at the N terminus.

Antibacterial Activities and the Positioning of Proline Residues—The antibiotic activities of the peptides were tested against two separate strains of *E. coli* and one strain of *P. aeruginosa* (Table 3). *E. coli* NCTC 9001 is a type strain, whereas TOP10 is a competent strain with a genotype similar to that of DH10B, which is deficient in *galU*, *galK*, and *galE* (34). Inactivation of *galE* perturbs the incorporation of glucose into the O-side chain of lipopolysaccharide (LPS) and hence *E. coli* TOP10 is expected to have an altered LPS structure. For each peptide, *E. coli* TOP10 was more susceptible to AMP challenge than the type strain. All of the designed peptides tested had potent activity against the Gram-negative bacteria used in the study with the activities of some peptides entering the nanomolar range, even for *P. aeruginosa*.

No significant differences in antibacterial activity were detected between the D-amino acid peptides. Importantly, however, small but statistically significant differences in activity could be detected between peptide analogs comprising L-amino acids. Notably, the proline-free peptides with small or intermediate charge angles, LAK80 and LAK120, performed poorly against *E. coli* NCTC 9001 when compared with the analog with the larger charge angle, LAK160. Within the LAK80 and LAK120 series, activity against *E. coli* NCTC 9001 was enhanced by the incorporation of proline ($p < 0.05$) irrespective of its position in the primary sequence. For the LAK120 series, a similar but less dramatic effect was seen when challenging *E. coli* TOP10, but when *P. aeruginosa* was chal-

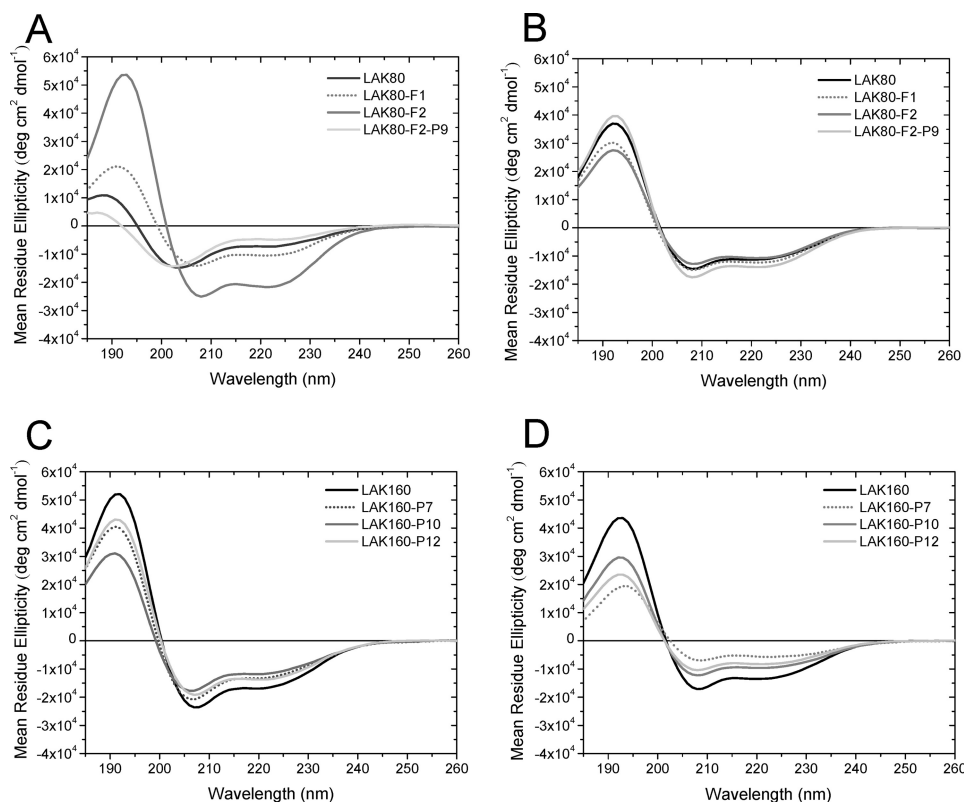


FIGURE 3. Circular dichroism spectra reveal the effect of peptide hydrophobicity and proline position on secondary structure in aqueous solution and membrane-mimicking medium. CD spectra are shown for LAK80, LAK80-F1, LAK80-F2, and LAK80-F2-P9 in 5 mM Tris buffer solution (A) or dimyristoyl phosphatidylcholine/dimyristoyl phosphatidylglycerol (75:25) liposomes (B) and LAK160, LAK160-P7, LAK160-P10, and LAK160-P12 in 50 mM SDS (C) or dimyristoyl phosphatidylcholine/dimyristoyl phosphatidylglycerol (75:25) liposomes (D). All spectra were recorded at 37 °C. deg, degrees.

TABLE 3

Comparison of MIC₅₀ values (μM) for 12 designed linear cationic peptides

Results are an average of two or more independently repeated experiments. ND, not determined. Proline-free template peptides are indicated in bold.

Peptide	<i>E. coli</i> NCTC 9001	<i>P. aeruginosa</i> PAO1	<i>E. coli</i> TOP10
LAK80	13.05 ± 0.13	ND	ND
LAK80-P7	2.97 ± 0.15	2.13 ± 0.77	1.26 ± 0.49
LAK80-P10	1.62 ± 0.17	1.18 ± 0.53	0.82 ± 0.03
LAK80-P12	1.80 ± 0.27	0.86 ± 0.09	0.80 ± 0.01
LAK120	2.95 ± 0.06	4.20 ± 0.92	1.47 ± 0.64
LAK120-P7	0.80 ± 0.03	4.45 ± 0.19	0.45 ± 0.10
LAK120-P10	0.78 ± 0.01	9.16 ± 1.43	0.47 ± 0.15
LAK120-P12	0.97 ± 0.06	2.14 ± 0.53	0.68 ± 0.14
LAK160	1.11 ± 0.48	2.23 ± 0.41	0.73 ± 0.40
LAK160-P7	0.79 ± 0.41	2.98 ± 0.21	0.46 ± 0.11
LAK160-P10	2.22 ± 0.63	4.14 ± 0.23	0.45 ± 0.24
LAK160-P12	0.48 ± 0.20	1.69 ± 0.09	0.38 ± 0.08

lenged, the positioning of the proline residue affected activity. In particular, the incorporation of proline in the LAK120 sequence was noted to increase activity ($p < 0.05$) when located at position 12 but decrease activity ($p < 0.05$) when located at position 10.

In contrast with LAK80 and LAK120, LAK160, the proline-free analog with the largest charge angle, had potent antimicrobial activity against each of the panel of Gram-negative organisms. Despite the high potency of the parent peptide, incorporation of proline at position 12 led to a modest but significant increase in activity against *E. coli* NCTC 9001 and *P. aeruginosa* ($p < 0.05$ and $p < 0.10$, respectively). Substitution of proline for leucine at position 7 had a less notable effect, but

performing the same substitution at position 10 led to a reduction in activity against the two strains (*E. coli* NCTC 9001, $p < 0.10$; *P. aeruginosa*, $p < 0.05$). This proline-containing analog, LAK160-P10, was therefore significantly ($p < 0.05$) less potent than its close relation, LAK160-P12, against both *E. coli* NCTC 9001 and *P. aeruginosa*. *E. coli* TOP10 was again more sensitive to each of the peptides in this series, and no significant differences in activity could be discerned. In summary, both the size of the charge angle and the positioning of the incorporated proline residues affected the activity against Gram-negative bacteria.

Because the LAK160 peptides showed little tendency to adopt α -helix-rich conformations in aqueous solution, the structure adopted in the target membrane environment may contribute to the significant differences in the observed antibacterial activity. The ability of the LAK160 peptides to adopt such conformations in the presence of membrane-mimicking anionic detergent SDS (Fig. 3C) or model dimyristoyl phosphatidylcholine/dimyristoyl phosphatidylglycerol (75:25) membranes (Fig. 3D) was confirmed. Micelles composed of the anionic detergent SDS provide a simple model for bacterial membranes, replicating the negative surface charge and hydrophobic/aqueous interface. In the presence of 50 mM SDS, all four LAK160 peptides adopted conformations with a high α -helix content with the positive and negative bands at 190, 207, and 220 nm of lower intensity for the proline-containing peptides as would be expected if these residues induced localized disruption of α -helix. However, the spectrum and result-

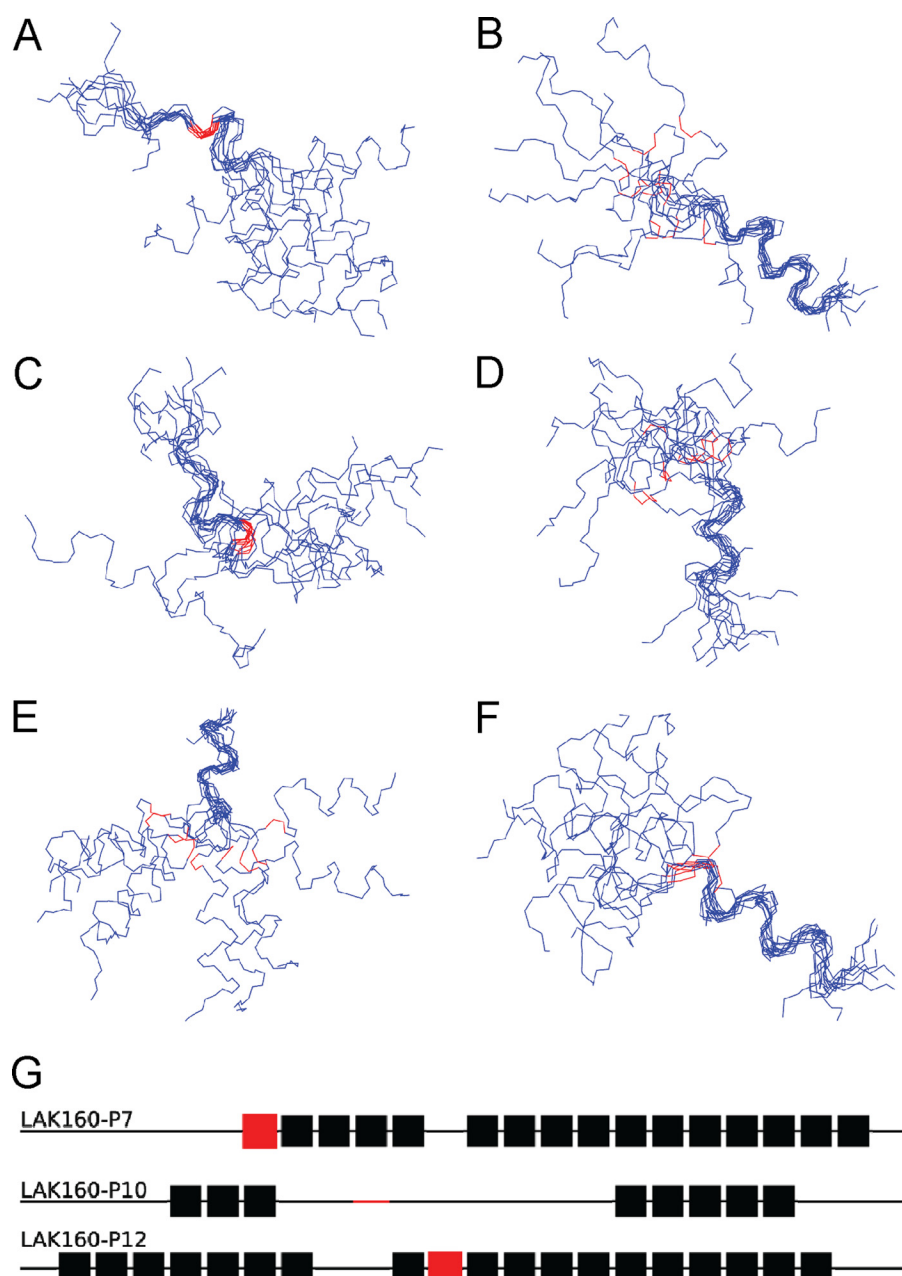


FIGURE 4. Shown is the superimposition of backbone atoms in each of the 10 lowest energy structures for LAK160-P7 (A and B), LAK160-P10 (C and D), and LAK160-P12 (E and F) fitted to either region 1 (A, C, and E) or region 2 (B, D, and F). Proline residues are shown in red. Also shown is a schematic representation of the 10 lowest energy structures where residues that are α -helical in more than eight structures are shown as solid boxes, and those that are α -helical in seven or fewer structures are shown as a line (G). Proline residues are again marked in red.

ant CDPro analysis (supplemental Fig. 7) of one peptide, LAK160-P10, indicated a substantially lower α -helix content in SDS but a higher content in model membranes when compared with the other proline-containing analogs. Because this peptide is modestly but significantly less potent against *E. coli* NCTC 9001 and *P. aeruginosa* than the other two LAK160 proline-containing analogs, we investigated the structural details of these three peptides further.

NMR Structure Determination—The NMR structures of the three proline-containing peptides (Fig. 4) confirm that the peptides adopt mostly α -helix conformation in the presence of 50 mM SDS with conformational flexibility induced through the incorporation of proline. Importantly, however, the amount of

α -helix conformation, the location and size of the region of flexibility, and the apparent role of the proline residues differ from one peptide to another within the LAK160-P series. The backbone atoms for each of the 10 lowest energy structures were superimposed, and it was found that the structures could be aligned to one of two separate helical regions in each peptide but not over the entire length. This was due to the conformational flexibility induced by the proline. The conformation adopted by the peptides is also shown schematically (Fig. 4G) where the number of structures adopting α -helix conformation at a given residue is used to reveal the predicted regions of high and low secondary structure. Notably, LAK160-P10 has two helical segments that are much smaller than those of either

LAK160-P7 or LAK160-P12; this is qualitatively in agreement with the NOESY (supplemental Fig. 8), NOE connectivities (supplemental Fig. 9), and CD spectra (Fig. 2C).

Interestingly, regardless of the position of the proline residue, all three LAK160 peptides form helices that are flexible near the middle of the sequence, although the effect of substituting proline for leucine in these three peptides may differ. Proline does not only act as a helix breaker; it also acts as a helix-stabilizing residue. Proline can stabilize an α -helix at its C terminus by taking part in a helix-capping interaction (35) in which the proline side chain caps the hydrophobic α -helix, and proline is the most water-soluble of all natural amino acid residues and therefore an ideal candidate for a position at the solvent-exposed α -helix end (35). Both LAK160-P7 and LAK160-P12 have two long α -helix regions separated by a short, one- or two-residue region that is predominantly unstructured. In contrast, for LAK160-P10, only three turns of α -helix are observed: one turn between Leu-5 and Leu-7 and two turns from Leu-17 to Leu-21. Furthermore, although residues Ala-8, Lys-9, Lys-15, Ala-16, and Lys-22 also have a tendency to form a helical structure as evidenced by their helical conformation in seven of the 10 lowest energy structures, LAK160-P12 exhibits a greater helical region running from Lys-2 to Lys-22 and is only interrupted at Lys-9 and Leu-10. Hence, even if the mentioned residues are included, LAK160-P10 is only capable of adopting a helical structure from Leu-5 to Lys-9 and from Lys-15 to Lys-22 with an unstructured region incorporating Leu-10 to Trp-14.

The prolines in LAK160-P7 and LAK160-P12 are in the expected position for such interactions to stabilize the N terminus of the α -helix. The helix in LAK160-P12 is longer compared with those in LAK160-P7 and LAK160-P10, which can be explained by the cooperative nature of α -helix formation. Because the first three residues at the N terminus can only form one hydrogen bond, a certain minimum number of amino acid residues capable of α -helix formation are needed. LAK160-P12 with proline at position 12 appears to satisfy this condition most effectively. If a well defined α -helix long axis is present, then the angle subtended by the charge angles can be measured, and the effect of proline substitution can be determined. The region of conformational flexibility identified in each peptide prevents identification of such an α -helix long axis. However, the structures do indicate that the amphipathic character of each peptide is retained with lysine residues segregated on one side of the structures (supplemental Fig. 10).

DISCUSSION

The prospects for the rational design of peptide antibiotics will be greatly enhanced by an understanding not only of how such peptides operate but also by how particular structural features contribute to a given mode of action. At present, a considerable body of work has focused on the interaction of cationic amphipathic α -helical peptides with bacterial membranes (36). Although many such peptides have been shown to disrupt bacterial membranes that are rich in anionic lipids, still others have been identified as having alternative modes of action, requiring penetration within the bacteria and inhibition of a number of processes that ultimately lead to bacterial cell death

(5). Furthermore, cationic antimicrobial peptides that have been shown to act against bacterial membranes may have multiple modes of action against different bacteria or indeed multiple effects against the same cell. The lipid composition of bacterial membranes varies considerably from species to species and has been shown to be a good predictor of the activity of some antimicrobial peptides (37). Peptides that are able to effectively cluster anionic lipids in the membranes of Gram-negative bacteria are particularly potent against such organisms (38–40). Magainin 2 does not effectively cluster anionic lipids most likely due to a low density of positive charges, but its membrane-disruptive behavior has been shown to depend on membrane composition (41), and it potentially has the ability to translocate into Gram-negative bacteria. Similarly, pleurocidin, a more potent cationic α -helical peptide against Gram-negative bacteria than magainin 2 (30) with strong membrane-disordering capabilities (42), has been reported to have the capacity to both form membrane pores (43) and inhibit intracellular processes (44). It is likely therefore that alterations in *e.g.* charge density in a peptide sequence may not only enhance this property at the expense of another but may also promote one mode of action at the expense of another.

Because it remains challenging to isolate the action of a peptide that ultimately leads to bacterial cell death and because a given peptide may have multiple effects on a bacterial cell, we are presently unable to predict how a given structural change in a peptide will influence antibacterial efficacy and mode of action. The aim of our work therefore was to precisely define the effects of structural alterations in a series of antimicrobial peptides on their biophysical properties and ultimately relate both of these to the efficacy and mode of action of the peptide. Here, we investigated the role of conformational flexibility conferred on the peptide through incorporation of the α -helix-breaking residue proline. Proline residues in natural antimicrobial peptides and modified sequences define a hinged region that is crucial for antibacterial potency and selectivity (11, 12, 45–52). In buforin II, the proline residue is proposed to operate as a translocation factor as the primary mode of antibacterial action for this peptide is considered to be the inhibition of intracellular functions (53). In a recent study, we have determined that although buforin has a high affinity for nucleic acids it adopts a low α -helix content in the presence of Gram-negative inner membrane-mimicking liposomes and is considerably less potent against such bacteria when compared with magainin and pleurocidin peptides (30). We were interested to see whether peptides that are designed to adopt conformations with high α -helix content but also incorporate proline residues can function as potent antibacterial agents.

In addition to causing a localized disruption of α -helix conformation, proline may have a number of effects on cationic α -helical antimicrobial peptide activity. These include the prevention of peptide self-association (12), leading to improved access through bacterial lipopolysaccharides to the bacterial inner membrane (54); the perturbation of the charged segment in the helical wheel, a key determinant of activity and selectivity (24); and the ability to translocate across the inner membrane, opening up intracellular modes of bactericidal action. Consistent with previous work (12), we have shown that proline has the

ability to affect peptide supramolecular organization and that this in turn is linked to greater potency against Gram-negative bacteria. Related observations have been made for other antimicrobial peptides: the cyclization of magainin and the substitution of amino acids of varying hydrophobicity on model membrane-active peptides led to reductions of antibacterial activity that were attributed to reduced flexibility and greater oligomerization, respectively (55, 56). Importantly, however, we have demonstrated that the positioning of the proline residue is crucial to the activity and that the structures adopted by the peptides in the presence of anionic SDS were defined by the location of the proline residue because this affected not only the location of the hinge region but also the conformation adopted in more remote regions of the sequence.

In addition, we have shown that the disruption of peptide structural ordering and its associated effects on supramolecular organization in solution are also linked to the antiplasmodial selectivity and ability to penetrate erythrocytes of all D-amino acid peptides and influence their cytotoxic activity against mammalian cancer cell lines. Understanding the antibacterial mode of action of AMPs has been the focus of a large body of work. Much less is known about how the same or related peptides act against eukaryotic targets such as *P. falciparum*. The present study showed that the incorporation of proline has a dramatic effect on the hemolytic potential of AMPs but that antiplasmodial activity is maintained. This indicates that the mechanism of AMP-induced hemolysis is distinct from that used to inhibit proliferation of *P. falciparum*. Furthermore, because the infected erythrocyte is not lysed and the parasite is effectively inhibited, the inference is that proline-containing peptides such as D-LAK120-AP13 or D-LAK120-HP13 are capable of penetrating erythrocytes to attack the intracellular pathogen. Proline-free peptides are incapable of this, and the observed antiplasmodial activity is difficult to separate from erythrocyte lysis. Although incorporation of proline in AMPs protects human erythrocytes from lysis, its presence enhances toxicity against nucleated cancer cells, again suggesting distinct mechanisms. The absence of cytotoxicity tests against primary human cells precludes any evaluation of the peptides as potential anticancer agents but nevertheless suggests that this feature may be beneficial. Related observations have shown that the lytic activity of anticancer peptides is inhibited by heparin sulfate on the surface of the target cells and that smaller peptides are better able to overcome this barrier (56, 58). Analogously, the incorporation of proline in larger AMPs and the concomitant prevention of self-association may aid their access to tumor cell membranes and enhance cytotoxic activity. Hence, although the extracellular barriers may differ from Gram-negative bacteria to *P. falciparum* and mammalian cells, the incorporation of proline may have a common effect in facilitating access to the target membrane.

In conclusion, our findings suggest that the environment-dependent structure adopted by antimicrobial peptides remains a determinant of activity against and selectivity toward a range of targets including Gram-negative bacteria, *P. falciparum*, and *M. tuberculosis* and that the main contribution of proline hinge regions is to modify the secondary structure and supramolecular assembly of the peptides in solution. This general feature

may facilitate access to target membranes and affect the mode of interaction on arrival. The three-dimensional structures of three proline-containing peptides show that significant changes in structure and activity are determined by the positioning of proline residues and that careful consideration of these effects is required to optimize potency against each target pathogen.

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