Chapter 7: Deformation & Strengthening Mechanisms

ISSUES TO ADDRESS...

- Why are the number of dislocations present greatest in metals?
- How are strength and dislocation motion related?
- Why does heating alter strength and other properties?



Dislocations & Materials Classes

- Metals (Cu, Al): Dislocation motion easiest
 - non-directional bonding
 - close-packed directions for slip
- Covalent Ceramics

 (Si, diamond): Motion difficult
 directional (angular) bonding
- Ionic Ceramics (NaCl): Motion difficult
 - need to avoid nearest
 - neighbors of like sign (- and +)









Dislocation Motion

Dislocation motion & plastic deformation

 Metals - plastic deformation occurs by slip – an edge dislocation (extra half-plane of atoms) slides over adjacent plane half-planes of atoms.



 If dislocations can't move, plastic deformation doesn't occur! Adapted from Fig. 7.1, *Callister & Rethwisch 8e.*



Dislocation Motion

- A dislocation moves along a slip plane in a slip direction perpendicular to the dislocation line
- The slip direction is the same as the Burgers vector direction



Lattice Strains Around Dislocations





Lattice Strain Interactions Between Dislocations





Deformation Mechanisms

Slip System

- Slip plane plane on which easiest slippage occurs
 - Highest planar densities (and large interplanar spacings)
- Slip directions directions of movement
 - Highest linear densities



 FCC Slip occurs on {111} planes (close-packed) in <110> directions (close-packed)

=> total of 12 slip systems in FCC

- For BCC & HCP there are other slip systems.

Stress and Dislocation Motion

• Resolved shear stress, τ_R

- results from applied tensile stresses



$\tau_{R} = \sigma \cos \lambda \cos \phi$



Critical Resolved Shear Stress



Single Crystal Slip



Adapted from Fig. 7.8, *Callister & Rethwisch 8e.*





Ex: Deformation of single crystal



So the applied stress of 45 MPa will not cause the crystal to yield.



Ex: Deformation of single crystal

What stress *is* necessary (i.e., what is the yield stress, σ_v)?

$$\tau_{crss} = 20.7 \text{ MPa} = \sigma_y \cos \lambda \cos \phi = \sigma_y (0.41)$$

$$\therefore \sigma_y = \frac{\tau_{crss}}{\cos \lambda \cos \phi} = \frac{20.7 \text{ MPa}}{0.41} = \underline{50.5 \text{ MPa}}$$

So for deformation to occur the applied stress must be greater than or equal to the yield stress

$$\sigma \ge \sigma_y = 50.5 \text{ MPa}$$



Slip Motion in Polycrystals

- Polycrystals stronger than single crystals - grain boundaries are barriers to dislocation motion.
- Slip planes & directions (λ, ϕ) change from one grain to another.
- τ_R will vary from one grain to another.
- The grain with the largest τ_R yields first.
- Other (less favorably oriented) grains yield later.



Adapted from Fig. 7.10, Callister & Rethwisch 8e. (Fig. 7.10 is courtesy of C. Brady, National Bureau of Standards [now the National Institute of Standards and Technology, Gaithersburg, MD].)



Strengthening Mechanisms: 1: Reduce Grain Size

- Grain boundaries are barriers to slip.
- Barrier "strength" increases with Increasing angle of misorientation.
- Smaller grain size: more barriers to slip.



Adapted from Fig. 7.14, *Callister & Rethwisch 8e.* (Fig. 7.14 is from *A Textbook of Materials Technology*, by Van Vlack, Pearson Education, Inc., Upper Saddle River, NJ.)

• Hall-Petch Equation:

$$\sigma_{yield} = \sigma_o + k_y d^{-1/2}$$



Strengthening Mechanisms : 2: Form Solid Solutions

- Impurity atoms distort the lattice & generate lattice strains.
- These strains can act as barriers to dislocation motion.
- Smaller substitutional impurity



Impurity generates local stress at **A** and **B** that opposes dislocation motion to the right.

 Larger substitutional impurity



Impurity generates local stress at **C** and **D** that opposes dislocation motion to the right.



Strengthening by Solid Solution Alloying

- Small impurities tend to concentrate at dislocations (regions of compressive strains) - partial cancellation of dislocation compressive strains and impurity atom tensile strains
- Reduce mobility of dislocations and increase strength



Adapted from Fig. 7.17, *Callister & Rethwisch 8e.*



Strengthening by Solid Solution Alloying

• Large impurities tend to concentrate at dislocations (regions of tensile strains)



(b)

Adapted from Fig. 7.18, *Callister & Rethwisch 8e.*



Ex: Solid Solution Strengthening in Copper

• Tensile strength & yield strength increase with wt% Ni.



• Alloying increases σ_y and *TS*.



Strengthening Mechanisms : 3: Cold Work (Strain Hardening)

- Deformation at room temperature (for most metals).
- Common forming operations reduce the cross-sectional area:



Dislocation Structures Change During Cold Working

• Dislocation structure in Ti after cold working.

0.2 µm



- Dislocations entangle with one another during cold work.
- Dislocation motion becomes more difficult.

Fig. 4.6, *Callister & Rethwisch 8e.* (Fig. 4.6 is courtesy of M.R. Plichta, Michigan Technological University.)



Impact of Cold Work

As cold work is increased

- Yield strength (σ_y) increases.
- Tensile strength (TS) increases.
- Ductility (%*EL* or %*AR*) decreases.



Mechanical Property Alterations Due to Cold Working

 What are the values of yield strength, tensile strength & ductility after cold working Cu?



%CW =
$$\frac{(15.2 \text{ mm})^2 - (12.2 \text{ mm})^2}{(15.2 \text{ mm})^2} \times 100 = \frac{35.6\%}{35.6\%}$$



Mechanical Property Alterations Due to Cold Working

• What are the values of yield strength, tensile strength & ductility for Cu for %CW = 35.6%?



Adapted from Fig. 7.19, *Callister & Rethwisch 8e.* (Fig. 7.19 is adapted from *Metals Handbook: Properties and Selection: Iron and Steels*, Vol. 1, 9th ed., B. Bardes (Ed.), American Society for Metals, 1978, p. 226; and *Metals Handbook: Properties and Selection: Nonferrous Alloys and Pure Metals*, Vol. 2, 9th ed., H. Baker (Managing Ed.), American Society for Metals, 1979, p. 276 and 327.) Chapter 7 -



Effect of Heat Treating After Cold Working

- 1 hour treatment at *T_{anneal}*... decreases *TS* and increases %*EL*.
- Effects of cold work are nullified!



- Three Annealing stages:
 - 1. Recovery
 - 2. Recrystallization
 - 3. Grain Growth

Adapted from Fig. 7.22, *Callister & Rethwisch 8e.* (Fig. 7.22 is adapted from G. Sachs and K.R. van Horn, *Practical Metallurgy, Applied Metallurgy, and the Industrial Processing of Ferrous and Nonferrous Metals and Alloys,* American Society for Metals, 1940, p. 139.)



Three Stages During Heat Treatment: 1. Recovery

Reduction of dislocation density by annihilation.

extra half-plane Scenario 1 of atoms **Dislocations Results from** annihilate atoms diffusion and form diffuse a perfect to regions atomic of tension plane. extra half-plane of atoms • Scenario 2 3. "Climbed" disl. can now ιR move on new slip plane 2. grey atoms leave by 4. opposite dislocations vacancy diffusion meet and annihilate allowing disl. to "climb" 1. dislocation blocked; Obstacle dislocation can't move to the right

Three Stages During Heat Treatment: 2. Recrystallization

- New grains are formed that:
 - -- have low dislocation densities
 - -- are small in size
 - -- consume and replace parent cold-worked grains.





Adapted from Fig. 7.21(a),(b), *Callister & Rethwisch 8e.* (Fig. 7.21(a),(b) are courtesy of J.E. Burke, General Electric Company.)



As Recrystallization Continues...

• All cold-worked grains are eventually consumed/replaced.



Adapted from Fig. 7.21(c),(d), *Callister & Rethwisch 8e.* (Fig. 7.21(c),(d) are courtesy of J.E. Burke, General Electric Company.)



Three Stages During Heat Treatment: 3. Grain Growth

- At longer times, average grain size increases.
 - -- Small grains shrink (and ultimately disappear)
 - -- Large grains continue to grow



Adapted from Fig. 7.21(d),(e), *Callister & Rethwisch 8e.* (Fig. 7.21(d),(e) are courtesy of J.E. Burke, General Electric Company.)

• Empirical Relation: exponent typ. ~ 2 grain diam. at time t. $d^n - d_o^n = Kt$ coefficient dependent
on material and *T*.
elapsed time





Recrystallization Temperature

 T_R = recrystallization temperature = temperature at which recrystallization just reaches completion in 1 h.

$$0.3T_m < T_R < 0.6T_m$$

For a specific metal/alloy, T_R depends on:

- %CW -- T_R decreases with increasing %CW
- Purity of metal -- T_R decreases with increasing purity



Diameter Reduction Procedure -Problem

A cylindrical rod of brass originally 10 mm (0.39 in) in diameter is to be cold worked by drawing. The circular cross section will be maintained during deformation. A cold-worked tensile strength in excess of 380 MPa (55,000 psi) and a ductility of at least 15 %*EL* are desired. Furthermore, the final diameter must be 7.5 mm (0.30 in). Explain how this may be accomplished.



Diameter Reduction Procedure -Solution

What are the consequences of directly drawing to the final diameter?



Diameter Reduction Procedure – Solution (Cont.)



- For %CW = 43.8%
 - $-\sigma_y = 420 \text{ MPa}$
 - *TS* = 540 MPa > 380 MPa
 - -% EL = 6 < 15
- This doesn't satisfy criteria... what other options are possible?



Adapted from Fig. 7.19,

Callister & Rethwisch 8e.

Diameter Reduction Procedure – Solution (cont.)



 \therefore our working range is limited to 12 < %CW < 27



Diameter Reduction Procedure – Solution (cont.)

Cold work, then anneal, then cold work again

- For objective we need a cold work of 12 < %CW < 27
 We'll use 20 %CW
- Diameter after first cold work stage (but before 2nd cold work stage) is calculated as follows:

$$\% CW = \left(1 - \frac{D_{f2}^{2}}{D_{02}^{2}}\right) \times 100 \implies 1 - \frac{D_{f2}^{2}}{D_{02}^{2}} = \frac{\% CW}{100}$$
$$\frac{D_{f2}}{D_{02}} = \left(1 - \frac{\% CW}{100}\right)^{0.5} \implies D_{02} = \frac{D_{f2}}{\left(1 - \frac{\% CW}{100}\right)^{0.5}}$$
Intermediate diameter = $D_{f1} = D_{02} = 7.5 \text{ mm} / \left(1 - \frac{20}{100}\right)^{0.5} = 8.39 \text{ mm}$

Diameter Reduction Procedure – Summary

Stage 1: Cold work – reduce diameter from 10 mm to 8.39 mm

%CW₁ =
$$\left(1 - \left(\frac{8.39 \text{ mm}}{10 \text{ mm}}\right)^2\right) x 100 = 29.6$$

Stage 2: Heat treat (allow recrystallization)

Stage 3: Cold work – reduce diameter from 8.39 mm to 7.5 mm

%CW₂ =
$$\left(1 - \left(\frac{7.5}{8.49}\right)^2\right) x 100 = 20$$
 Fig 7.19
 \Rightarrow $TS = 400$ MPa

Therefore, all criteria satisfied

% EL = 24



Cold Working vs. Hot Working

• Hot working \rightarrow deformation above T_R

• Cold working \rightarrow deformation below T_R



Summary

- Dislocations are observed primarily in metals and alloys.
- Strength is increased by making dislocation motion difficult.
- Strength of metals may be increased by:
 - -- decreasing grain size
 - -- solid solution strengthening
 - -- precipitate hardening
 - -- cold working
- A cold-worked metal that is heat treated may experience recovery, recrystallization, and grain growth – its properties will be altered.

